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2013 HSC Chemistry - Board of Studies

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Sample answer: Hydrochloric acid is a strong acid and therefore fully ionises in aqueous solution. Ethanoic acid is a weak acid and does not fully ionise in aqueous solution. The total concentration of ethanoic acid must be higher than that of HCl, due to incomplete

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Target Publications Pvt. Ltd. Chemistry board Answer paper: march 2013 Chemistry SECTION - I Q.1. Select and write the most appropriate answer from the given alternatives for each sub-question: i. (A) 68% [1] ii. (C) atm s-1 [1] iii. (B) Thiosulphuric acid [1] iv.

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