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least according to the physics textbook, that is.) E1-26 The room volume is $(21 \times 13 \times 12)\text{ft}^3(0.3048 \text{ m/ft})^3 = 92.8\text{m}^3$. The mass contained in the room is $(92.8\text{m}^3)(1.21\text{kg/m}^3) = 112\text{kg}$: E1-27 One mole of sugar cubes would have a volume of $N_A \times 1.0 \text{ cm}^3$, where N_A is the Avogadro constant.

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